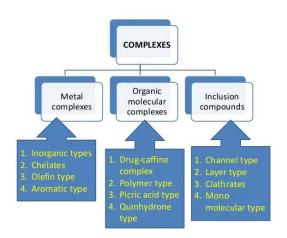
COMPLEXATION:

DEFINATION: Complexation is the process of complex formation that is the process of characterization the covalent or non-covalent interactions between two or more compounds.

CLASSIFICATION OF COMPLEXATION:



1) METAL ION OR CO- ORDINATION COMPLEXES

A) INORGANIC COMPLEXES:

In Inorganic metal complexes, ligands provides only one site for binding with metals. **EX** - Carbohydrates, fats, protein, nucleic acid, urea .

B) CHELATES: Chelates are a group of metal ion complexes in which substances provides two or more dinars group to combined with metal ion.

EX- Ethylene diamine.

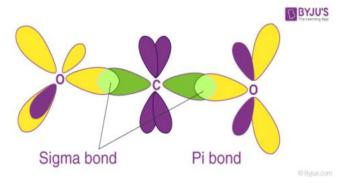
C) OLEFIN TYPE: These are used as catalyst in the manufacturing of the bulk drugs and analysis of drugs.

Ex - Unsaturated hydrocarbons

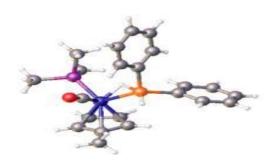
D) Aromatic type:

a) Pie complex — (pi = Cd)

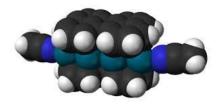
EX – Aromatic bases (benzene, toluene and xylene) form pi-bond complexes with metal ions like Ag by Lewis acid base reaction.



b) Sigma complex: Sigma bond complexes involve in the formation of a sigma bond between ion and carbon of aromatic ring.



c) Sandwich Complexes: They are relatively stable complexes involving in the delocalized covalent bond between the d-orbital of transition metal and a molecular orbit of the aromatic ring.



2) ORGANIC MOLECULAR COMPLEX

A) Quinhydrone type: (C12H10O4) The molecular complex of this type of is obtained by mixing alcoholic solutions of equimolar quantities of hydroquinone and

benzoquinone **EX** – benzene ,hydrogen , Quinone.

B) Picric acid: (C6H3N3O7)It is pale yellow, odorless crystal that is slightly soluble in water. Ex- Bateson Picrate.

C) Caffeine and other drug complexes:

(C8H10N4O2) A number of acidic drugs are known to form complexes with caffeine. Drugs such as benzocaine, procaine and tetra Caine form complexes with caffeine.

D) Polymer type:

Many pharmaceutical additives such as polyethylene glycols(PEGs), carboxymethyl cellulose(CMC) contain nucleophilic oxygen.

3) Inclusion or occlusion compounds:

A) lattice type:

Channel lattice complexes provides a mean of separation of optical isomers.

B)Lattice type:

Layer type of complex is the is the type of inclusion compounds in which guest molecules is diffused between the layer of carbon atoms, to form a alternate guest and host molecules.

C) Clathrates:

The clathrates are compounds that crystallize in the form of a cage like lattices in which the coordinating compounds is entrapped.

D) Monomolecular type:

Monomolecular inclusion complexes provides entrapment of guest molecules into the cage like structure formed from a single host molecules.

E) Macromolecular type

The atoms are arranged in three dimensions to produce cages and channels in which the guest molecules are entrapped.

APPLICATION OF COMPLEXATION:

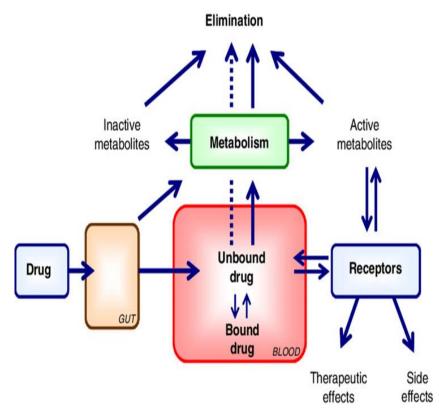
COMPLEXATION – Applications in pharmacy

- •Physical state
- Volatility
- Solid state Stability
- Chemical stability
- Solubility
- Dissolution

- Partition coefficient
- ·Absorption & bioavailability
- •Reduced toxicity
- •Antidote in metal poisoning
- •Drug action through metal poisoning
- Antibacterial activity

PROTEIN BINDING:

- 1) protein binding refers to the degree to which medications attach to proteins within the blood.
- 2) drug's efficiency may be affected by the degree to which it binds.



1)Elimination – It means the act, process, or an instance of eliminating or discharging.

- 2)Inactive Metabolites When a metabolites of a drugs produces no therapeutic effect it is considered an inactive metabolites.
- 3) Active Metabolites When a drug is metabolized by the body into a modified form which continues to produce effects in the body.
- 4)Metabolism The metabolic breakdown of drugs by living organisms, usually through specialized enzymatic systems.
- **5)Drug** A chemical substance used to treat, cure, prevent, or diagnose a disease or to promote well-being.
- **6)Receptors** —Receptors are large molecules usually proteins, that interact with and mediate the action of drugs.

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