HETEROCYCLIC COMPOUNDS

1. PYRAZOLE

Molecular formula– $C_3H_4N_2$ Molecular weight – 68.07 gm/mol

Structure -



Resonance structure-



Synthesis-

Knorr Pyrazole Synthesis -



Chemical properties -

| Reactant | Reagent | Product |
|--------------|-------------------|-----------------|
| ESR | | |
| Sulphonation | Sulphuric Acid | Pyrazole |
| Pyrazole | | Sulphonic Acid |
| Halogenation | Chlorine | Chloro Pyrazole |
| Pyrazole | | |
| Nitration | Nitric acid, | Nitro Pyrazole |
| Pyrazole | Sulphuric Acid | |
| Reduction | H2(Hydrogenation) | Pyrazolidine |
| Pyrazole | | |
| Oxidation | O3(Ozonolysis) | Glyoxal |
| Pyrazole | | |

Medicinal Uses -



Used as Anti cancer (Pyrazofurin)
Used as Anti Inflammatory(Lanazolac)
Used as Analgesics(Difenamizole)

- Used as Analgesics(Difendinizor
- •Used as Vasodilator(Sildenafil)
- •Used as anti depressant (Fezolamide)

2. IMIDAZOLE

Molecular Formula –C₃H₄N₂ Molecular Weight –68.077gm/mol

Structure -



Resonance Structure –



Synthesis – By action of NH₃ on glyoxal –



Chemical properties -

| Reactant | Reagent | Product |
|--------------|----------------|------------------|
| ESR | | |
| Nitration | Nitric Acid | 4-Nitroimidazole |
| Imidazole | | |
| Sulphonation | Sulphuric Acid | 4-Imidazole |
| Imidazole | | Sulphonic Acid |
| Halogenation | Chlorine | 4- |
| Imidazole | | Chloroimidazole |
| Alkylation | Potassium | 1- |
| Imidazole | Hydroxide, | methylimidazole |
| | Iodomethane | |
| Oxidation | Hydrogen | Oxamide |
| Imidazole | peroxide | |



3. OXAZOLE

Molecular formula $-C_3H_3NO$

Molecular Weight- 69.06 gm/mol

Structure -



Resonance structure -



Synthesis -

Fischer oxazole Synthesis -



Chemical properties -

| Rectant | Reagent | Product |
|---------------|-------------|---------------------|
| ESR | | |
| Nitration | Nitric acid | 5-nitrazole |
| Oxazole | | |
| Sulphonation | Sulphuric | 5-oxazole sulphonic |
| Oxazole | acid | acid |
| Halogenation | Chlorine | 5-Chlorazole |
| Oxazole | | |
| Oxidation | Manganese | Imide |
| Oxazole | dioxide | |
| Deprotonation | Lithium | Isonitride |
| Oxazole | | |



4. THIOZOLE

Molecular Formula - C₃H₃NS Molecular Weight - 85.13gm/mol

Structure -



Resonance Structure -



Synthesis -

Hantzsch Thiazole Synthesis



Chemical Properties -

| Reactant | Reagent | Product |
|--------------|-------------------|------------------------|
| ESR | | |
| Nitration | Nitric acid | 5- nitrothiazole |
| Thiazole | | |
| Halogenation | Chlorine | 5 – chloro thiazole |
| Thiazole | | |
| Sulphonation | Sulphuric acid | 5 – thiazole sulphonic |
| Thiazole | | acid |
| Oxidation | Hydrogen peroxide | Thiazole-N-Oxide |
| Thiazole | | |



5. PYRIDINE

Molecular formula – C₅H₅N Molecular Weight – 79.1 gm/mol

Structure –



Resonance structure -



Synthesis -

Hatzsch pyridine synthesis -



Chemical properties –

| Reactant | Reagent | Product |
|--------------|-----------------|--------------------|
| ESR | | |
| Nitration | Nitric acid | Nitropyridine |
| Pyridine | | |
| Halogenation | Chlorine | Chloropyridine |
| Pyridine | | |
| Sulphonation | Sulphuric acid | Pyridine sulphonic |
| Pyridine | | acid |
| Reduction | H2 | Piperidine |
| Pyridine | (hydrogenation) | |



6. QUINOLINE

Molecular formula – C₉H₇N

Molecular weight – 129.16 gm/mol

Structure-



Resonance structure -



Synthesis – Skraup quinoline sunthesis –



Chemical properties -

| Reactant | Reagent | Product |
|---------------------------|----------------------------------|-------------------------|
| ESR | | |
| Nitraion | Nitric acid | 5-nitroquinoline |
| Quinoline | Sulphuric acid | 5 quincline |
| Quinoline | | sulphonic acid |
| Halogenation Quinoline | Bromine | 5-bromoquinoline |
| Oxidation Quinoline | CH ₃ CCOO-OH/peracids | Quinoline-n-oxide |
| Reduction Quinoline | H2/Pt (hydrogenation) | Tetrahydro quinoline |



7. ISOQUINOLINE

 $Molecular\ formula - C_9 H_7 N$

Molecular weight -129.16 gm/mol

Structure -



Resonance structure -



Synthesis –

From cinnamaldehyde –



Chemical properties -

| Reactant | Reagent | Product |
|--------------|-----------------------|---------------------|
| ESR | | |
| Nitration | Nitric acid | 5-nitroisoquinoline |
| Isoquinoline | | |
| Halogenation | Bromine | 5-bromoisoquinoline |
| Isoquinoline | | |
| Reduction | H ₂ /Pt/Ni | Decahydro |
| Isoquinoline | (Hydrogenation) | isoquinoline |
| Oxidation | RCOO-OH | N-oxide |
| Isoquinoline | Peracids | |



8. ACRIDINE

Molecular formula – C₁₃H₉N

Molecular weight – 179.13 gm/mol

Structure -



Synthesis -From diphenyl amine





Diphenyl amine

Acridine

Chemical properties –

| Reactant | Reagent | Product |
|-------------------------------------|------------------------------|--|
| ESR | | |
| Nitration Acridine | Nitric acid | 2-nitro acridine |
| Halogenation Acridine | Chlorine/Bromine | 2-chloro acridine/ 2-bromo acridine |
| NSR With soda amide Acridine | NaNH ₂ | 9-aminoacridine |
| Reduction Acridine | Pt/HCl | Octahydroacridine |
| Reductive alkylation Acridine | n-pentonoic acid UV light | 9-butylacridine |



9. INDOLE

Molecular formula $- C_8 H_7 N$

Molecular weight – 117.15 gm/mol

Structure -

Resonance structure –



Synthesis -

Fischer indole synthesis –



Chemical properties -

| Reactant | Reagent | Product |
|------------------------|---|--------------------------|
| ESR | | |
| Halogentaion Indole | Chlorine | Chloroindole |
| Nitration Indole | Nitric acid | Nitroindole |
| Reducion Indole | Dil.acid | 3H-indolium cation |
| Oxidation Indole | Mn-MC6*a H ₂ O ₂ | 3-hydroxy- indolenine |



10.PYRIMIDINE

 $Molecular\ formula - {C_4}{H_4}{N_2}$

Structure -



Synthesis -

From 1, 3 dicarbonyl compounds

Medicinal uses -

11.PURINE

Molecular formula $- C_5H_4N_4$

Structure -

Synthesis -

From uric acid

12. AZEPINES

$Molecular\ formula - C_{6}H_{7}N$

Structure -

Synthesis -

From nitrobenzene

Medicinal uses -

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- 2. Textbook of Pharmaceutical Organic Chemistry By PV Publication
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